



UTTRAKHAND OPEN UNIVERSITY, HALDWANI (NAINITAL)

M.Sc. Second Year Chemistry*Last Date of Submission:*

15 May, 2015

Course Title: Reaction Mechanism, Pericyclic Reaction, Photochemistry and Stereochemistry**Course Code: CHE551****Year: 2014-15****Maximum Marks: 40 Marks****Section 'A'**

Section 'A' contains 08 short answer type questions of 5 marks each. Learners are required to answer 4 questions only. Answers of short answer-type questions must be restricted to 250 words approximately.

1. Explain Norrish type-I and Norrish type II reactions with suitable examples.
2. Explain E2 mechanism with a suitable example.
3. How are the free radicals generated from peroxide, nitrite and a carboxylate anion? Formulate the mechanism of bromination of cyclohexene by NBS.
4. Write a short note on:
 - a. Carbenoid
 - b. Triplet carbene
 - c. Hofmann rule
5. Discuss the mechanism of the followings:
 - a. Baeyer-Villiger rearrangement
 - b. Pinacol-Pinacolone rearrangement
6. Define cycloaddition reaction, What are [m+n] cycloaddition? Explain with suitable examples.
7. Draw the π MO of allylic carbocation. What are the HOMO and LUMO in its ground state and first excited state?
8. Explain the following with suitable examples:
 - a. Kinetic isotope reaction
 - b. Deconjugation
 - c. Phosphorescence

Section 'B'

Section 'B' contains 04 long answer-type questions of 10 marks each. Learners are required to answer 02 questions only.

1. a. Explain the stereochemistry of [3,3] shift by PMO method.
b. Discuss the boat conformations of cyclohexane. Why the boat conformation of cyclohexane is less stable than the chair conformation.
2. Write a short note on:
 - i. Di- π Methane rearrangement
 - ii. Paterno-Buchi reaction
 - iii. Fries Rearrangement Reaction
 - iv. Lossen Rearrangement
3. Write down the mechanism of the following name reactions.
 - a. Curtius Rearrangement
 - b. Lossen Rearrangement
 - c. Hofmann Rearrangement
4. Discuss the stability of carbanions. Formulate the mechanism of an addition reaction of a carbanion.

